

Available online at www.sciencedirect.com

Tetrahedron 60 (2004) 3493–3503

Tetrahedron

Synthesis of 3-hydroxyindolin-2-one alkaloids, (\pm) -donaxaridine and (\pm) -convolutamydines A and E, through enolization–Claisen rearrangement of 2-allyloxyindolin-3-ones

Tomomi Kawasaki,* Miyuki Nagaoka, Tomoko Satoh, Ayako Okamoto, Rie Ukon and Atsuyo Ogawa

Meiji Pharmaceutical University, 2-522-1 Noshio, Kiyose, Tokyo 204-8588, Japan

Received 26 November 2003; accepted 9 February 2004

Abstract—Claisen rearrangement triggered by enolization of 2-allyloxyindolin-3-ones with DBU was performed in order to prepare 3-allyl-3-hydroxyindolin-2-ones. Total synthesis of 3-hydroxyindolin-2-one alkaloids, (\pm) -donaxaridine, as well as (\pm) -convolutamydines A and E, was achieved by transformation of the allyl moiety of 3-allyl-3-hydroxyindolin-2-ones. $©$ 2004 Elsevier Ltd. All rights reserved.

1. Introduction

3-Substituted 3-hydroxyindolin-2-ones are useful synthetic intermediates for alkaloids and biologically active compounds such as donaxaridine (1) (1) (1) ,¹ convolutamydines (2) (2) (2) ,² $dioxibrassinine, ³$ $dioxibrassinine, ³$ $dioxibrassinine, ³$ welwitindolinone C,^{[4](#page-9-0)} 3'-hydroxygluco-isatisin,^{[5](#page-9-0)} and TMC-95s, 6 in addition to several others ([Fig. 1](#page-1-0)) \cdot ^{[7](#page-9-0)} In particular, 3-allyl-3-hydroxyindolin-2-ones are attractive intermediates for synthesis of biologically active compounds. Although a number of routes to 3-substituted 3-hydroxyindolin-2-ones are known, $8-17$ there are relatively few synthetic methods for 3-allyl-3-hydroxyindolin-2-ones. The known examples are addition of allylmetallic (indium,^{[18](#page-10-0)} gallium^{[19](#page-10-0)} and boran^{[20](#page-10-0)}) reagents to isatin, but there are difficulties in obtaining the desired 3-allyl-3 hydroxyindolin-2-one owing to the low regioselectivity of the allylic reaction site.^{[18,19](#page-10-0)} Reaction of allylmagnesium chloride with isatin resulted in diallylation to give only 2,2- diallylindolin-3-one.^{[21](#page-10-0)} Recently, Mérour et al.²¹ reported alkaline hydrolysis of 2-ethoxycarbonyl-2-allyloxyindolin-3-ones followed by decarboxylation and Claisen rearrangement to give 3-allyl-3-hydroxyindolin-2-one. We have previously shown a synthetic methodology for regioselective introduction of an allyl moiety to an indole nucleus using Claisen rearrangement, converting 3-allyl-oxyindole to 2-allylindolin-3-one,^{[22](#page-10-0)} 3-alkyl-2-allyloxyin-dole to 3-alkyl-3-allylindolin-2-one^{[23](#page-10-0)} and 3-vinyloxyindoline to 4-carbamoylmethylindoles[.24](#page-10-0) We herein report a method

for synthesis of the 3-hydroxyindolin-2-one alkaloids, (\pm) donaxaridine (1) as well as (\pm) -convolutamydines A (2a) and E (2b), using Claisen rearrangement triggered by enolization of 2-allyloxyindolin-3-ones 3 to 3-allyl-3-hydroxyindolin-2 ones 4 ([Scheme 1](#page-1-0)).

2. Results and discussion

2.1. Preparation of 3-allyl-3-hydroxyindolin-2-ones

The starting 2-allyloxyindolin-3-ones 3 were readily available using our synthetic method.^{[23](#page-10-0)} Initially, we examined the enolization of 2-allyloxyindolin-3-one 3a with DBU and DBN as a base under several reaction conditions and the results are summarized in [Table 1.](#page-1-0) When 3a was treated with DBU at 40 \degree C in acetonitrile, the desired enolization readily took place through Claisen rearrangement of an intermediary indole to afford 3-allyl-3-hydroxyindolin-2-one 4a, its deacetyl derivative 5a, and carboxylic acid 6 in 9, 12, and 36% yields, respectively ([Scheme 2](#page-1-0) and [Table 1,](#page-1-0) entry 1). It is known that Claisen rearrangement of the enolate of α -allyloxy carbonyl compounds competes with [2,3]-Wittig rearrangement.^{[25](#page-10-0)} However, comparison of the ${}^{13}C$ NMR spectrum of 5a with those of 3-hydroxyindolin-2-one 7 and 2-hydroxyindolin-3 one 8[26](#page-10-0) shows that the product in the reaction of 4a is not the [2,3]-Wittig rearrangement product $5a'$ but the Claisen rearrangement product 5a ([Fig. 2\)](#page-2-0). Formation of 5a and 6 is effected by hydrolysis of 4a under basic reaction conditions, and thus 4a was smoothly hydrolyzed with lithium hydroxide at room temperature to give 5a in 85% yield. When the reaction was performed in methylene chloride

Keywords: DBU; Allylalcohol; Osmium tetraoxide; Wacker oxidation; Molybdenum peroxide.

 $*$ Corresponding author. Fax: $+81-424-95-8763$;

e-mail address: kawasaki@my-pharm.ac.jp

Figure 1. 3-Hydroxy-indolin-2-one alkaloids.

Scheme 1. Enolization–Claisen rearrangement.

Scheme 2. Enolization–Claisen rearrangement of 2-allyloxyindolin-3-one 3a.

Table 1. Reaction conditions in enolization–Claisen rearrangement of 2-allyloxyin, rdolin-3-one 3a

Entry	Base	Solvent	Reaction temperature $(^{\circ}C)$	Reaction time (min)	Yield $(\%)$		
					4a	5a	o
	DBU	MeCN	40	30	$\mathbf Q$	12	36
2	DBU	CH_2Cl_2	40	10	12	16	30
3	DBU	Toluene	40	20	70	16	
4	DBU	Toluene	rt	80	56	13	
5	DBU	Toluene	110			55	3
6	DBN	Toluene	40	10		70	
	DBN	Toluene	rt	60		63	6

instead of acetonitrile, the reaction was completed in a shorter time, though by-product 6 was still formed (entry 2). Using toluene as the reaction solvent resulted in reduced formation of 6, thus improving the total yields of 4a and 5a (entry 3). Further attempts to carry out the reaction at various temperatures (entries 4 and 5) confirmed that the reaction conditions shown in entry 3 were the most suitable. When DBN was used instead of DBU, the reaction proceeded more smoothly to give the desired 5a as the

chief product in good yield; however, small amounts of byproduct 6 were formed (entries 6 and 7).

Next, we investigated the DBU-promoted reaction of various 2-allyloxyindolin-3-ones 3b-f. The results are summarized in [Scheme 3](#page-2-0) and [Table 2](#page-2-0). When 3b was treated with DBU in toluene at 40° C for 10 min, 3-(2-methyl-2buten-2-yl)-3-hydroxyindolin-2-one 4b and the deacetylated 5b were obtained in 70 and 9% yields (entry 1). The

Figure 2. 13C NMR spectra of hydroxyindolin-2-ones and 2-hydroxyindolin-3-one.

reaction of 3c under the same conditions required prolonged heating (120 min) to give the Claisen products 4c and 5c in moderate yields (entry 2). A similar reaction of (E) cinnamyl derivative 3d for 20 min afforded a mixture of the diastereoisomers of 4d (50%, 1.5:1) together with 5d (20%) (entry 3). As examples of secondary rather than primary ethers, reactions of 2-buten-2-yl and cyclo-2 hexenyl derivatives 3e and 3f were performed. The reaction of 3e proceeded through stereoselective Claisen rearrangement to give (E) -buten-2-ylindolin-3-ones 4e (54%) and 5e (8%) (entry 4). The (E) -product 4e is predominantly produced via chair-like transition state A which is more favorable than boat-like transition state \bf{B} [\(Fig. 3](#page-3-0)).^{[25](#page-10-0)} For the reaction of 3f, 4f and 5f were obtained as respective mixtures of their diastereoisomers (4:1) in 59% and 18% yields (entry 5).

Scheme 3. Enolization–Claisen rearrangement of 2 allyloxyindolin 3-one 3b-f.

Table 2. Preparation of 3-allyl-3-hydroxyindolin-2-ones 4 and 5

^a The ratio of diastereomers measured by HPLC; **4d** (1:1.4), **5d** (1.1:1), **4f** (2:1), **5f** (4:1).

Figure 3. Transition states A and B in Claisen rearrangement of 3e.

2.2. Synthesis of (\pm) -donaxaridine and (\pm) convolutamydines A and E

For the synthesis of 3-hydroxyindolin-2-one alkaloids, donaxaridine (1) and convolutamydines A $(2a)$ and E (2b), we attempted transformation of the allyl group of 3-allyl-3-hydroxyindolin-2-one 5a to aldehyde and acetonyl groups. When $\overline{5a}$ was treated with OsO₄ and N-methylmorpholine N -oxide (NMO) followed by NaIO₄, the unstable aldehyde was obtained and used in the following reaction without purification. Reductive amination of the aldehyde with N a BH ₃ CN in the presence of methylammonium chloride gave the 3-methylaminoethyl-3 hydroxyindolin-2-one (\pm) -donaxaridine (1) in 41% overall yield from 5a (Scheme 4). All spectral data are identical to those of the natural and synthetic samples.^{[1](#page-9-0)}

For transformation of the allyl group to the acetonyl group, we utilized Wacker oxidation because Wacker oxidation of olefins containing b-oxygenated functional groups with O_2 -PdCl₂-CuCl₂ or –CuCl in DMF–H₂O regioselectively afforded the corresponding ketones.^{[27](#page-10-0)} However, Wacker oxidation of 5a and its O-TBDMS derivative 9 under these conditions was very slow (Scheme 5 and Table 3, entries 1–5). When 9 was allowed to react with O_2 -PdCl₂-CuCl in 1,4-dioxane-H₂O, the desired Wacker oxidation readily took place to afford 10 in high yield (entry 6). In addition, the reaction using other Pd catalysts was attempted to give moderate results (entries 7 and 8). Deprotection of 10 with tetrabutylammonium fluoride (TBAF) and acetic acid gave the 3-acetonyl-3-hydroxyindolin-2-one 11 in moderate yield.[28](#page-10-0)

Finally, we applied these preparative methods to the total synthesis of (\pm) -convolutamydines A (2a) and E (2b). 4,6-Dibromoindolin-3-one 14 was readily obtained from 4,6-dibromoindole 12^{29} 12^{29} 12^{29} by molybdenum peroxide oxidation followed by demethoxylation^{[30](#page-10-0)} in 50% overall

Scheme 4. Reagents: (a) OsO₄, NMO, MeCN, rt, 2 h and then NalO₄, dioxane–H₂O (2:1), rt, 10 min. (b) MeNH₂·HCl, MeOH, rt, 3 h and then NaBH₃CN, rt, 36 h, 41% (overall yield from 5a).

Scheme 5. Reagents: (a) TBDMSOTf, 2,6-lutidine, CH₂Cl₂, 0 °C to rt, 3 days, 73%. (b) See Table 3. (c) TBAF, AcOH, 0 °C to rt, 50%.

Table 3. Wacker oxidation of 3-allylindolin-2-ones **5a** and 9 with $O₂$

Entry	Olefin	Reagents (equiv.)	Solvents	Temperature $(^{\circ}C)$	Time	Yield $(\%)$
	5a	$PdCl2-CuCl2(0.1:1)$	$DMF-H2Oa$	rt	week	
		$PdCl2-CuCl2(0.1:1)$	$DMF-H2Oa$	rt	week	
		$PdCl2-CuCl2(1:10)$	$DMF-H2Oa$	rt	week	10
4		$PdCl2-CuCl2(1:10)$	$DMF-H2Oa$	50	1 week	
		$PdCl2-CuCl$ (0.1:1)	$DMF-H2Oa$	50	4 days	
6		$PdCl2-CuCl$ (0.1:1)	$dioxane-H2Oa$	50	.5 h	88
		$PdCl2(PhCN)2-CuCl2(0.1:1)$	$DMF-H2Oa$	50	dav	25
8		$Na_2PdCl_2-t-BuO_2H$ (0.2:1.5)	$AcOH-H2Ob$	50	5 h	71

^a The ratio of solvents (7:1).
^b The ratio of solvents (1:1).

yield. Successive bromination of 14, substitution with allyl alcohol, DBU-promoted enolization–Claisen rearrangement of 15 and hydrolysis with LiOH gave the desired 4,6-dibromo-3-hydroxyindolin-2-one 16 in 84% yield (Scheme 6). The $OsO₄ - NaIO₄$ oxidation of 16 followed by NaBH₄ reduction yielded (\pm) -convolutamydine E (2b) in 65% yield (Scheme 7).

Synthesis of (\pm) -convolutamydine A (2a) was achieved by TBDMS protection of 16 followed by Wacker oxidation of 17^{[31](#page-10-0)} and deprotection of 18 with tris(dimethylamino)sulfur (trimethylsilyl)difluoride (TAS-F) (Scheme 7).^{[32](#page-10-0)} All spectral data of 2a and 2b are identical to those of the natural and synthetic samples.^{[2](#page-9-0)}

3. Conclusion

In conclusion, we have presented a general and useful method for synthesis of 3-allyl-3-hydroxyindolin-2-ones 5 using Claisen rearrangement triggered by DBU-promoted enolization of 2-allyloxyindolin-3-ones 3. As examples of the synthetic utility of 5, we performed transformation of the allyl group in 5a to aldehyde and acetonyl groups, and respectively applied these methodologies to achieve total

synthesis of (\pm) -donaxaridine (1) as well as (\pm) -convolutamydines $A(2a)$ and $E(2b)$.

4. Experimental

4.1. General

¹H NMR spectra were obtained using a JEOL JNM-EX-300, JNM-EX-400, or JNM-LA-500 spectrometer with tetramethylsilane as an internal standard. J-Values are given in Hz. Mass spectra were obtained using a JEOL JMS-DX302 or JMS-700 instrument with a direct inlet system operating at 70 eV. IR spectra were recorded with a Shimadzu FTIR-8100 spectrophotometer. All mp values are uncorrected, and were measured on a Yanagimoto micromelting point apparatus. HPLC was performed on a JASCO PU-1580 with a JASCO Finepak SIL-5 column. Elemental analyses were obtained using a Yanaco CHN Corder MT-6 elemental analyzer. Column chromatography was carried out on silica gel (Kanto Chemical Co. Inc., Silica Gel 60N, 100–200 mesh and Merck, Silica Gel 60, 230–400 mesh). Preparative TLC was undertaken using Merck Silica Gel 60 F_{254} .

Scheme 6. Reagents: (a) MoO₅·HMPA, MeOH, rt, 1 week, 90%. (b) CSA, MeCN, reflux, 56%. (c) Br₂, CH₂Cl₂, 0 °C, for 2h, and then allyl alcohol, MS 4A, DMF, rt, for 2 days. (d) DBU, toluene, 40 °C, 3 h, and then LiOH, MeOH, rt, 3 days, 84% (overall yield from 14).

Scheme 7. Reagents: (a) OsO₄, NMO, MeCN, rt 1 h, and then NaIO₄, dioxane–H₂O, rt, 1 h. (b) NaBH₄ MeOH, 0 °C, 30 min., 65% (c) TBDMSOTf, 2,6lutidine, 0 °C to rt, 3 days, and then AcOH, H₂O, THF, 80 °C, 2 h, 69%. (d) PdCl₂, CuCl, dioxane–H₂O (7:1), 50 °C, 24 h, 12%, (e) TAS–F, 0–15 °C, 2.5 h, 75%.

4.2. General procedure for the preparation of 3-allyl-3 hydroxyindolin-2-ones 4-5

A solution of 2-allyloxyindolin-3-ones 3 (1.0 mmol) and DBU or DBN (1.0 mmol) in dry solvent (13 ml) as shown in [Table 1](#page-1-0) was stirred at a designated temperature (rt \sim 110 °C) under nitrogen atmosphere for a designated period (10–120 min), as shown in [Tables 1 and 2.](#page-1-0) The reaction mixture was neutralized with AcOH at 0° C and extracted with EtOAc. The extract was washed with H_2O , dried over MgSO4, and concentrated under reduced pressure to give a residue. The residue was subjected to chromatography on a silica gel column with $EtOAc$ –hexane $(1:1-2)$ as an eluent to give 3-allyl-3-hydroxyindolin-2-ones 4, deacetyl derivatives 5, and carboxylic acid 6.

4.2.1. 1-Acetyl-3-allyl-3-hydroxyindolin-2-one (4a). Mp 98-101 °C (EtOAc-hexane). IR (CHCl₃) cm⁻¹: 3559, 1763, 1717. ¹H NMR (CDCl₃, 300 MHz) δ: 2.59 (3H, s, $-$ Ac), 2.60 (1H, dd, J=13.1, 8.8 Hz, $-CH₂$), 2.68 (1H, dd, J=13.1, 6.2 Hz, -CH₂-), 2.80 (1H, br, -OH), 5.07 (1H, d, $J=11.2$ Hz, $-CH=CH_2$), 5.08 (1H, d, $J=16.0$ Hz, $-CH=CH_2$), 5.49 (1H, m, $-CH=CH_2$), 7.19 (1H, td, $J=7.5$, 1.0 Hz, Ar-H), 7.32 (1H, td, $J=7.5$, 1.5 Hz, Ar-H), 7.37 (1H, ddd, $J=7.5$, 1.5, 0.6 Hz, Ar-H), 8.15 (1H, d, $J=7.5$ Hz, Ar-H). MS m/z (%): 231 (M⁺, 11), 190 (47), 162 (22), 148 (100), 43 (11). HRMS m/z calcd for $C_{13}H_{13}NO_3$: 231.0895. Found: 231.0896. Anal. Calcd for $C_{13}H_{13}NO_3$: C, 67.52; H, 5.67; N, 6.06. Found: C, 67.39; H, 5.72; N, 5.82.

4.2.2. 3-Allyl-3-hydroxyindolin-2-one (5a). Mp 112– 115 °C (EtOAc–hexane); [lit.²⁰ 123–124 °C]. IR (CHCl₃) cm⁻¹: 1728, 1624. ¹H NMR (CDCl₃, 400 MHz) δ: 2.61 (1H, dd, J=13.2, 8.5 Hz, –CH₂–), 2.75 (1H, dd, J=13.2, 6.3 Hz, $-CH_2$ –), 3.65 (1H, brs, $-OH$), 5.10 (1H, d, J=18.1 Hz, $-CH=CH_2$), 5.11 (1H, d, J=8.8 Hz, $-CH=CH_2$), 5.65 (1H, m, $-CH=CH_2$), 6.88 (1H, d, J=7.6 Hz, Ar-H), 7.07 (1H, t, $J=7.6$ Hz, Ar-H), 7.25 (1H, t, $J=7.3$ Hz, Ar-H), 7.36 (1H, d, $J=7.3$ Hz, Ar-H). ¹³C NMR (CDCl₃, 100 MHz) δ : 42.9, 76.3, 110.3, 120.4, 122.9, 124.3, 129.5, 130.06, 130.14, 140.1, 180.0. MS m/z (%): 189 (M⁺, 9), 148 (100), 120 (3), 92 (3), 65 (3), 39 (2). HRMS m/z calcd for $C_{11}H_{11}NO_2$: 189.0790. Found: 189.0788.

4.2.3. 1-Acetyl-3-(2-methyl-3-buten-2-yl)-3-hydroxyindolin-2-one (4b). Mp $73-75$ °C (EtOAc–hexane). IR $(CHCl₃)$ cm⁻¹: 3565, 1765, 1713. ¹H NMR (CDCl₃, 300 MHz) ^d: 1.06 (3H, s, C–Me), 1.17 (3H, s, C–Me), 2.61 (3H, s, –Ac), 2.84 (1H, brs, –OH), 5.14 (1H, dd, $J=17.6$, 1.1 Hz, $-CH=CH_2$), 5.22 (1H, dd, $J=10.8$, 1.1 Hz, $-CH=CH_2$), 5.94 (1H, dd, $J=17.6$, 10.8 Hz, $-CH = CH_2$), 7.21 (1H, td, J=7.5, 1.1 Hz, Ar-H), 7.36 $(1H, td, J=7.7, 1.5 Hz, Ar-H), 7.42 (1H, dd, J=7.5, 1.5 Hz,$ Ar-H), 8.21 (1H, dt, J=8.4, 0.6 Hz, Ar-H). MS m/z (%): 259 $(M⁺, 0.2), 217 (0.2), 191 (100), 162 (9), 149 (80), 69 (55),$ 41 (14). HRMS m/z calcd for $C_{15}H_{17}NO_3$: 259.1208. Found: 259.1208.

4.2.4. 3-(2-Methyl-3-buten-2-yl)-3-hydroxyindolin-2-one (5b). Mp $188-190$ °C (EtOAc–hexane). IR (CHCl₃) cm⁻¹: 3431, 1732. ¹H NMR (CDCl₃, 300 MHz) δ : 1.12 (3H, s, C–Me), 1.18 (3H, s, C–Me), 2.84 (1H, brs, –OH), 5.15 (1H, dd, $J=17.4$, 1.2 Hz, $-CH=CH_2$), 5.24 (1H, dd, $J=10.8$, 1.2 Hz, $-CH=CH_2$), 6.19 (1H, dd, $J=17.4$, 10.8 Hz, $-CH=CH_2$), 6.81 (1H, d, J=7.6 Hz, Ar-H), 7.02 $(1H, td, J=7.6, 0.9 Hz, Ar-H), 7.25 (1H, td, J=7.6 1.3 Hz,$ Ar-H), 7.37 (1H, dd, J=7.6, 0.7 Hz, Ar-H). MS m/z (%): 217 $(M⁺, 2)$, 149 (100), 119 (3), 69 (15), 41 (7). HRMS m/z calcd for $C_{13}H_{15}NO_2$: 217.1103. Found: 217.1098. Anal. Calcd for $C_{13}H_{15}NO_2$: C, 71.87; H, 6.96; N, 6.45. Found: C, 72.05; H, 6.99; N, 6.17.

4.2.5. 1-Acetyl-3-(2-methyl-2-propenyl)-3-hydroxyindo**lin-2-one (4c).** Viscous oil. IR $(\text{CHCl}_3) \text{ cm}^{-1}$: 3559, 1765, 1713. ¹H NMR (CDCl₃, 300 MHz) δ: 1.40 (3H, s, $-CMe=CH_2$), 2.60 (3H, s, $-Ac$), 2.66 (1H, d, $J=12.9$ Hz, $-CH_2$, 2.75 (1H, dd, $J=12.9$, 0.7 Hz, $-CH_2$, 2.98 (1H, br, $-OH$), 4.63 (1H, d, $J=0.9$ Hz, $-CMe=CH_2$), 4.80 (1H, t, J=1.7 Hz, –CMe=CH₂), 7.24 $(1H, td, J=7.4, 1.1 Hz, Ar-H), 7.37 (1H, td, J=7.4, 1.5 Hz,$ Ar-H), 7.42 (1H, ddd, J=7.4, 1.5, 0.6 Hz, Ar-H), 8.18 (1H, d, J=7.7 Hz, Ar-H). ¹³C NMR (CDCl₃, 100 MHz) δ : 23.8, 26.5, 47.2, 76.4, 116.5, 116.6, 123.9, 125.4, 128.8, 130.1, 138.3, 139.6, 170.3, 178.5. MS m/z (%): 245 (M⁺, 22), 190 (42), 162 (31), 148 (100), 130 (3), 102 (3), 90 (4), 43 (10). HRMS m/z calcd for $C_{14}H_{15}NO_3$: 245.1052. Found: 245.1052.

4.2.6. 3-(2-Methyl-2-propenyl)-3-hydroxyindolin-2-one (5c). Mp 163 °C (EtOAc–hexane). IR (CHCl₃) cm⁻¹: 3435, 1736. ¹H NMR (CDCl₃, 300 MHz) δ: 1.54 (3H, s, $-CMe=CH_2$), 1.60 (1H, br, -OH), 2.68 (2H, s, -CH₂-), 4.65 (1H, d, J=1.1 Hz, –CMe=CH₂), 4.78 (1H, t, $J=1.7$ Hz, $-CMe=CH_2$), 6.83 (1H, d, $J=7.5$ Hz, Ar-H), 7.05 (1H, td, $J=7.5$, 0.9 Hz, Ar-H), 7.24 (1H, td, $J=7.5$, 1.3 Hz, Ar-H), 7.35 (1H, d, $J=7.5$ Hz, Ar-H), 7.76 (1H, br, $-NH$). MS m/z (%): 203 (M⁺, 11), 185 (4), 148 (100), 119 (5), 92 (3), 65 (3). HRMS m/z calcd for $C_{12}H_{13}NO_2$: 203.0946. Found: 203.0945. Anal. Calcd for $C_{12}H_{13}NO_2$: C, 70.92; H, 6.45; N, 6.89. Found: C, 70.74; H, 6.54; N, 6.55.

4.2.7. 1-Acetyl-3-(1-phenyl-2-propenyl)-3-hydroxyindo**lin-2-one** (4d). Viscous oil. IR $(CHCl₃)$ cm⁻¹: 3550, 1765, 1715. ¹H NMR (CDCl₃, 300 MHz) δ: 2.31 (3H×0.6, s, -Ac), 2.55 (3H×0.4, s, -Ac), 2.97 (1H×0.6, br, -OH), 3.18 (1H \times 0.4, br, -OH), 3.82 (1H \times 0.4, d, J=10.5 Hz, $-CH-Ph$), 3.86 (1H \times 0.6, d, J=8.3 Hz, $-CH-Ph$), 5.26 $(1H \times 0.6, dt, J=16.9, 1.3 Hz, -CH=CH_2)$, 5.39 (1H, ddd, $J=10.5$, 2.4, 1.3 Hz, $-CH=CH_2$), 5.43 (1H \times 0.4, dd, $J=16.9$, 1.3 Hz, $-CH=CH_2$), 6.30 (1H \times 0.4, m, $-CH=CH_2$), 6.37 (1H \times 0.6, m, $-CH=CH_2$), 6.70 (1H, m, Ar-H), 6.9–7.5 (7H, m, Ph, Ar-H), 7.97 (1H \times 0.4, d, J=8.1 Hz, Ar-H), 7.99 (1H×0.6, d, J=7.9 Hz, Ar-H). MS m/z (%): 307 (M⁺, 0.3), 289 (0.5), 247 (0.6), 148 (12), 117 (100), 91 (4). HRMS m/z calcd for C₁₉H₁₇NO₃: 307.1208. Found: 307.1205.

The ratio (1:1.4) of two diastereoisomers was determined by HPLC.

4.2.8. 3-(1-Phenyl-2-propenyl)-3-hydroxyindolin-2-one (5d). Mp 179–[18](#page-10-0)1 °C (EtOAc–hexane) [lit.¹⁸ mp 160– 162 °C; 1:1 diastereomer mixture]. IR (CHCl₃) cm⁻¹: 3420, 1736. ¹H NMR (CDCl₃, 300 MHz) δ: 2.79 (1H, brs, -OH), 3.85 (1H×0.4, d, J=9.7 Hz, -CH-Ph), 3.88 (1H×0.6, d, $J=8.1$ Hz, $-CH-Ph$), 5.25 (1H \times 0.4, dt, $J=17.0$, 1.4 Hz, $-CH=CH_{2}$), 5.3–5.4 (1H, m, $-CH=CH_{2}$), 5.44 (1H \times 0.6, dd, $J=17.0$, 1.4 Hz, $-CH=CH_2$), 6.3–6.55 (1H, m, $-CH = CH_2$), 6.63 (1H, m, Ar-H), 6.8–7.55 (7H, m, Ar-H). MS m/z (%): 265 (M⁺, 2), 247 (2), 148 (38), 117 (100), 91 (7). HRMS m/z calcd for $C_{17}H_{15}NO_2$: 265.1103. Found: 265.1095.

The ratio (1.1:1) of two diastereoisomers was determined by HPLC.

4.2.9. (E)-1-Acetyl-3-(2-butenyl)-3-hydroxyindolin-2-one (4e). Mp 50–55 °C (EtOAc–hexane). IR (CHCl₃) cm⁻¹: 3561, 1765, 1713. ¹H NMR (CDCl₃, 300 MHz) δ : 1.60 (3H, d, $J=6.3$ Hz, $-CH=CHMe$), 2.61 (3H, s, $-Ac$), 2.5–2.69 $(2H, m, -CH_2-CH=CH)$, 2.97 (1H, br, $-OH$), 5.19 (1H, dddd, $J=15.2$, 8.7, 6.3, 1.7 Hz, –CH=CHMe), 5.56 (1H, dq, $J=15.2$, 6.3 Hz, –CH=CHMe), 7.23 (1H, td, $J=7.5$, 1.1 Hz, Ar-H), 7.36 (1H, td, $J=7.5$, 1.8 Hz, Ar-H), 7.42 $(1H, ddd, J=7.5, 1.5, 0.6 Hz, Ar-H), 8.18 (1H, d, J=8.2 Hz,$ Ar-H). MS mlz (%): 245 (M⁺, 12), 203 (3), 191 (43), 162 (32), 148 (100), 130 (3), 102 (3), 90 (4), 43(11). HRMS m/z calcd for $C_{14}H_{15}NO_3$: 245.1051. Found: 245.1053.

4.2.10. (E)-3-(2-Butenyl)-3-hydroxyindolin-2-one (5e). Mp $115-120$ °C (EtOAc-hexane). IR (CHCl₃) cm⁻¹: 3435, 1726. ¹H NMR (CDCl₃, 300 MHz) δ : 1.61 (3H, d, $J=6.4$ Hz, $-CH=CHMe$), 2.52 (1H, dd, $J=13.5$, 8.5 Hz, $-CH_2$ -CH=CH), 2.63 (1H, dd, J=13.5, 6.4 Hz, $-CH_2$ - $CH=CH$), 5.31 (1H, ddd, $J=15.0$, 8.4, 6.4 Hz, $-CH = CHMe$, 5.63 (1H, dq, $J=15.0$, 6.4 Hz, $-CH = CHMe$), 6.84 (1H, d, J=7.6 Hz, Ar-H), 7.06 (1H, td, $J=7.6$, 0.9 Hz, Ar-H), 7.25 (1H, td, $J=7.6$, 1.3 Hz, Ar-H), 7.34 (1H, d, $J=7.6$ Hz, Ar-H), 7.67 (1H, br, $-NH$). $MS \, m/z \, (\%): 203 \, (M^+, 12), 185 \, (7), 170 \, (8), 148 \, (100), 119$ (3), 92 (4), 65 (3). HRMS m/z calcd for $C_{12}H_{13}NO_2$: 203.0946. Found: 203.0943.

4.2.11. 1-Acetyl-3-(2-cyclohexenyl)-3-hydroxyindolin-2 one (4f). Mp $158-160$ °C (EtOAc–hexane). IR (CHCl₃) cm⁻¹: 3550, 1738. ¹H NMR (CDCl₃, 300 MHz) δ : 0.7–2.0 $(6H, m, -(CH₂)₃ -), 2.58$ and 2.59 (3H, s, -COMe), 2.77 $(H, m, -CH-CH₂), 5.6-5.9$ (2H, m, $-CH=CH-$), 7.1– 7.2 (1H, m, Ar-H), 7.25–7.4 (2H, m, Ar-H), 8.17 (1H, d, $J=8.3$ Hz, Ar-H). MS m/z (%): 271 (M⁺, 1), 253 (9), 211 (12), 191 (100), 149 (69), 81 (73), 43 (8). HRMS m/z calcd for $C_{16}H_{17}NO_3$: 271.1208. Found: 271.1210.

The ratio (2:1) of two diastereoisomers was determined by HPLC.

4.2.12. 3-(2-Cyclohexenyl)-3-hydroxyindolin-2-one (5f). Mp $158-163$ °C (EtOAc-hexane). IR (CHCl₃) cm⁻¹: 3440, 1728. ¹H NMR (CDCl₃, 300 MHz) δ : 0.87 (1H, m, –CH), 1.4–2.0 (5H, m, –(CH₂)₃–), 2.80 (1H, m, –CH– CH_2 -), 5.6–5.8 (2H \times 0.2, m, –CH=CH–), 5.95 (1H \times 0.8, m, $-CH=CH-$), 6.07 (1H \times 0.8, m, $-CH=CH-$), 6.85 (1H, d, $J=7.5$ Hz, Ar-H), 7.04 (1H \times 0.8, td, $J=7.5$, 0.9 Hz, Ar-H), 7.07 (1H \times 0.2, td, J=7.5, 0.9 Hz, Ar-H), 7.3–7.45 (2H, m, Ar-H). MS m/z (%): 229 (M⁺, 3), 149 (100), 119 (3) , 92 (3), 81 (23), 65 (3). HRMS m/z calcd for $C_{14}H_{15}NO_2$: 229.1103. Found: 229.1094. Anal. Calcd for $C_{14}H_{15}NO_2$: C, 73.34; H, 6.59; N, 6.11. Found: C, 73.04; H, 6.34; N, 5.88. The ratio (4:1) of two diastereoisomers was determined by HPLC.

4.2.13. 2-[2-(Acetylamino)phenyl]-2-hydroxypent-4 enoic acid (6). Viscous oil. IR $(CHCI₃)$ cm⁻¹: 3437, 3368, 1740, 1624. ¹H NMR (CDCl₃, 300MHz) δ : 2.07 $(3H, s, -COMe), 2.61$ (1H, dd, J=13.6, 7.9 Hz, $-CH_2$ -CH=), 2.81 (1H, ddt, J=13.6, 6.4, 1.2 Hz, $-CH_2$ -CH=), 5.09 (1H, dq, $J=17.2$, 1.0 Hz, $-CH=CH_2$), 5.10 (1H, dq, $J=9.5$, 1.0 Hz, $-CH=CH_2$), 5.62 (1H, dddd, $J=16.3$, 11.0, 8.3, 6.6 Hz, $-CH=CH_2$), 6.85 (1H, d, J=7.7 Hz, Ar-H), 7.03 (1H, td, J=7.5, 0.9 Hz, Ar-H), 7.21 (1H, d, J=7.7 Hz, Ar-H), 7.26 (1H, td, $J=7.7$, 1.3 Hz, Ar-H). ¹³C NMR (CDCl3, 75 MHz) ^d: 22.6, 40.9, 79.4, 110. 0, 120.7, 122.6, 123.3, 127.6, 129.2, 129.7, 140.6, 169.0, 171.2, 175.6. MS m/z (%): 213 (M⁺-H₂O, 23), 190 (13), 148 (100). HRMS $(M⁺-H₂O)$ m/z calcd for C₁₃H₁₃NO₃: 231.0896. Found: 231.0900.

4.3. Synthesis of (\pm) -donaxaridine (1)

4.3.1. Hydrolysis of 1-acetyl-3-allyl-3-hydroxyindolin-2 one (4a) to 3-allyl-3-hydroxyindolin-2-one (5a). A solution of $4a$ (51 mg, 0.2 mmol) and LiOH (10%, 0.1 ml) in MeOH (2 ml) was stirred at room temperature for 75 min. The resulting mixture was evaporated under reduced pressure to give a residue, which was extracted with EtOAc. The extract was washed with water, dried over MgSO4, and concentrated under reduced pressure to give a residue. The residue was purified by column chromatography on silica gel with EtOAc–hexane (2:1) as an eluent to give 5a (45 mg, 85%).

4.3.2. 3-(2-Methylaminoethyl)-3-hydroxyindolin-2-one $[(\pm)$ -donaxaridine, 1]. A solution of 5a (89 mg, 0.47 mmol), $OsO₄$ (4% in water, 182 µl, d=4.9, 0.14 mmol), and NMO (50% in water, 194 μ l, $d=1.13$, 0.94 mmol) in MeCN (3.5 ml) was stirred at room temperature for 2 h. The reaction mixture was filtered through Celite® 545. The filtrate was concentrated under reduced pressure to give a residue containing 1,2-diol. A solution of the residue and $NaIO₄$ (100 mg, 0.78 mmol) in 1,4-dioxane–H₂O $(1.5 \text{ ml}, 2.1)$ was stirred at room temperature for 10 min. After diluting the resulting mixture with diethyl ether, the organic layer was washed with H_2O , and the aqueous layer was extracted with EtOAc. The combined organic solution was dried over MgSO4, and concentrated under reduced pressure to give the crude aldehyde 7 [IR (CHCl₃) cm⁻¹; 3435, 1738, 1728. ¹H NMR $(CDCl_3, 300 MHz)$ δ : 2.95 (1H, dd, J=16.6, 1.7 Hz, $-CH_2CHO$), 3.02 (1H, dd, $J=16.6$, 1.7 Hz, $-CH_2CHO$), 4.24 (1H, br, OH), 6.89 (1H, d, $J=7.7$ Hz, Ar-H), 7.05 (1H, t, $J=7.7$ Hz, Ar-H), 7.25 (1H, t, $J=7.7$ Hz, Ar-H), 7.32 (1H, d, $J=7.7$ Hz, Ar-H), 8.54 (1H, br, NH), 9.80 (1H, t, $J=1.7$ Hz, $-CH₂CHO$], which was used without further purification because of its instability.

A solution of 7 and MeNH₂·HCl (56 mg, 0.85 mmol) in MeOH (2.7 ml) was stirred at room temperature. After disappearance of 7 was confirmed by TLC $(3 h)$, NaBH₃CN (82 mg, 1.3 mmol) was added to the reaction mixture at room temperature. The mixture was stirred for 36 h and extracted with CH_2Cl_2 . The extract was washed with H_2O ,

dried over MgSO₄, and concentrated under reduced pressure to give a residue. The residue was subjected to chromatography on a silica gel column with EtOAc–hexane (2: 1) as an eluent to give (\pm) -donaxaridine (1, 40 mg, 41%). Mp 175 °C; [lit.^{[1a](#page-9-0)} 175–176 °C]. IR (KBr) cm⁻¹; 3404, 3236, 1673, 1613, 1470, 1308. ¹H NMR (CDCl₃, 300 MHz) δ: 2.42 (1H, td, J=12.8, 9.0 Hz, $-C-CH_2-CH_2$), 2.77 (1H, ddd, $J=12.8$, 6.2, 1.7 Hz, $-C-CH_2-CH_2$), 2.97 (3H, s, $-NMe$), 3.26 (1H, td, J=9.5, 6.2 Hz, $-CH_2-N-$), 3.34 (1H, ddd, $J=9.5$, 9.5, 1.7 Hz, $-CH_2-N-$), 4.31 (1H, brs, $-OH$), 4.71 (1H, br, –NH), 6.69 (1H, td, $J=7.6$, 1.3 Hz, Ar-H), 6.72 (1H, dd, J=7.9, 0.7 Hz, Ar-H), 6.88 (1H, dd, J=7.6, 1.5 Hz, Ar-H), 7.11 (1H, td, $J=7.6$, 1.5 Hz, Ar-H). MS m/z $(\%): 206 \ (M^+, 100), 188 \ (9), 177 \ (8), 173 \ (7), 159 \ (6), 149$ (42), 148 (11), 147 (29), 146 (37), 135 (24), 130 (19), 120 (58), 93 (13), 92 (18), 77 (6), 65 (15), 58 (43). HRMS m/z calcd for $C_{11}H_{14}N_2O_2$:206.1055. Found: 206.1050.

4.4. Synthesis of 3-acetonyl-3-hydroxyindolin-2-one (11)

4.4.1. 3-Allyl-3-tert-butyldimethylsilyloxyindolin-2-one (9). A solution of 3-hydroxyindolin-2-one 5a (57 mg, 0.3 mmol), 2,6-lutidine (128 mg, 1.2 mmol), and tertbutyldimethylsilyl triflate (317 mg, 1.2 mmol) in dry CH_2Cl_2 (4 ml) was stirred at 0 °C to room temperature for 1 h. The reaction mixture was diluted with CH_2Cl_2 . The organic layer was washed with brine, dried over $MgSO₄$, and evaporated to give a residue, of which a solution in THF–AcOH–H₂O (1:1:1, 1 ml) was stirred at 80 °C for 5 h. The resulting mixture was diluted with $CH₂Cl₂$. The organic layer was washed with brine, dried over MgSO₄, and concentrated under reduced pressure to give a residue, which was purified by silica gel column chromatography with EtOAc–hexane (1:3) as an eluent to give TBDMS ether 9 (67 mg, 73%). viscous oil. IR (CHCl₃) cm⁻¹: 1720.
¹H NMR (300 MHz, CDCl₂) δ : -0 26 (3H_s, -SiMe), 0.06 ¹H NMR (300 MHz, CDCl₃) δ : -0.26 (3H, s, -SiMe), 0.06 $(3H, s, -SiMe), 0.87 (9H, s, -Si'Bu), 2.53 (1H, dd, J=13.4,$ 8.2 Hz, $-CH_2CH=$), 2.74 (1H, dd, J=13.4, 6.4 Hz, $-CH_2CH=$), 5.02 (2H, m, $-CH=CH_2$), 5.66 (1H, m, $-CH=CH₂$), 6.86 (1H, d, J=7.7 Hz, Ar-H), 7.03 (1H, dd, J=7.7, 8.3 Hz, Ar-H), 7.25 (2H, m, Ar-H), 8.74 (1H, brs, $-NH$). ¹³C NMR (CDCl₃, 100 MHz) δ : -3.9, -3.5, 18.2, 25.8, 44.3, 78.0, 110.2, 119.1, 122.2, 124.6, 129.2, 130.9, 131.0, 140.0, 180.0. MS (FAB) m/z (%): 304 (MH⁺, 11), 246 (18), 205 (19), 172 (100). HRMS (FAB, MH⁺) m/z calcd for $C_{17}H_{26}NO_2Si: 304.1733$. Found: 304.1739.

4.4.2. Typical procedure for Wacker oxidation of 3 allylindolin-2-one 9 to 3-(2-oxopropyl)-3-TBDMS-oxyindolin-2-one (10). A suspension of palladium (II) chloride (8 mg, 0.0043 mmol) and copper (I) chloride (4.3 mg, 0.043 mmol) in dioxane– H_2O (7:1, 1 ml) was vigorously stirred with bubbling oxygen gas at room temperature for 1 h. The indolin-2-one 9 (13 mg, 0.043 mmol) was added to the mixture. After heating at 50 °C for 5 h, the reaction mixture was cooled to room temperature and diluted with $CH₂Cl₂$. The precipitate was filtered off, and the filtrate was dried over MgSO4, and evaporated off. The obtained residue was purified by flash silica gel column chromatography with EtOAc–hexane (2:3) as an eluent to give 3-(2-oxopropyl) indolin-2-one 10 (12 mg, 88%). viscous oil. ¹H NMR (300 MHz, CDCl₃) δ : -0.19 (3H, s, -SiMe), 0.08 (3H, s, -SiMe), 0.95 (9H, s, -Si^tBu), 2.14 (3H, s, -COMe), 3.07

(1H, d, J=15.9 Hz, $-CH_2CO-$), 3.25 (1H, d, J=15.9 Hz, $-CH_2CO-$), 6.83 (1H, d, J=7.8 Hz, Ar-H), 7.02 (1H, t, J=7.8 Hz, Ar-H), 7.21-7.29 (2H, m, Ar-H), 7.46 (1H, brs, $-NH$). MS (FAB) m/z (%): 320 (MH⁺, 54), 262 (37), 188 (38), 146 (100), 73 (44). HRMS (FAB, MH⁺) m/z calcd for $C_{17}H_{25}NO_3Si: 320.1682.$ Found: 320.1671.

4.4.3. 3-(2-Oxopropyl)-3-hydroxyindolin-2-one (11). A solution of TBDMS derivative 10 (8.4 mg, 0.026 mmol), TBAF (1.0 M, 22 μ l, 22 μ mol), and AcOH (15 μ l) in THF (1 ml) was stirred at room temperature for 3 days. After addition of EtOAc, the organic layer was washed with brine, dried over MgSO4, and concentrated under reduced pressure. The residue was purified by preparative TLC with EtOAc as a developing solvent to give 3-hydroxy-indolin-2-one 11 (2.5 mg, 50%). Mp 165–168 °C [lit.^{[9a](#page-9-0)} mp 166–168 °C]. IR (CHCl₃) cm⁻¹: 3445, 1740, 1624. ¹H NMR (300 MHz, CDCl₃) δ : 2.20 (3H, s, –COMe), 2.97 (1H, d, J=17.0 Hz, $-CH_2CO-$), 3.19 (1H, d, J=17.0 Hz, $-CH_2CO$ –), 4.37 (1H, br, –OH), 6.87 (1H, d, J=7.5 Hz, Ar-H), 7.05 (1H, td, J=7.5, 0.9 Hz, Ar-H), 7.27 (1H, td, $J=7.5$, 1.2 Hz, Ar-H), 7.35 (1H, d, $J=7.5$ Hz, Ar-H), 7.65 (1H, br, -NH). MS m/z (%): 205 (M⁺, 88), 187 (24), 172 (27), 162 (98), 148 (100), 120 (72), 92 (41), 43 (36). HRMS m/z calcd for $C_{11}H_{11}NO_3$: 205.0739. Found: 205.0742.

4.5. Syntheses of (\pm) -convolutamydines A (2a) and E (2b)

4.5.1. 1-Acetyl-4,6-dibromo-3-hydroxy-2-methoxy-indoline (13). A solution of 1-acetyl-4,6-dibromoindole^{[29](#page-10-0)} (12, 7.0 g, 22.0 mmol) and hexamethylphosphoramideoxodiperoxomolybdenum (VI) $(MoO₅·HMPA)$ (17.2 g 46.3 mmol) in MeOH (450 ml) was stirred at room temperature for a week. The resulting mixture was concentrated under reduced pressure, treated with sat. $Na₂SO₃$, and extracted with $CH₂Cl₂$. The extract was washed with H_2O , dried over $MgSO_4$, and concentrated under reduced pressure. The obtained residue was purified by silica gel column chromatography with EtOAc–hexane (1:1) as an eluent to give 3-hydroxy-2-methoxyindoline 13 (7.2 g, 90%). Mp 165–168 °C (CH₂Cl₂). IR (CHCl₃) cm⁻¹: 3600, 1686. ¹H NMR ?300 MHz, CDCl₃) δ : 2.32 (3H, s, N–COMe), 2.37 (1H, brs, OH), 3.43 (3H, s, –OMe), 4.89 (1H, brs, –CH–OH), 5.28 (1H, brs, –CH–OMe), 7.46 (1H, d, $J=1.7$ Hz, Ar-H), 8.27 (1H, br, Ar-H). MS m/z (%): 367 $(M+4, 25)$, 365 $(M+2, 53)$, 363 $(M⁺, 26)$, 324 (47), 322 (89), 320 (43), 292 (62), 290 (100), 288 (46), 280 (26), 278 (21), 263 (16), 213 (10), 211 (16), 149 (13), 43 (59). HRMS m/z calcd for $C_{11}H_{11}Br_2NO_3$: 362.9106. Found: 362.9106.

4.5.2. 1-Acetyl-4,6-dibromoindolin-3-one (14). A solution of 13 (364 mg, 1.0 mmol) and 10-camphorsulfonic acid (715 mg, 3.08 mmol) in MeCN was heated under reflux for 30 min. The reaction mixture was evaporated under reduced pressure and extracted with EtOAc. The extract was washed with sat. NaHCO₃ and brine, dried over $MgSO₄$, and concentrated under reduce pressure. The residue was purified by silica gel column chromatography with EtOAc–hexane (1:1) as an eluent to give indolin-3-one 14 (186 mg, 56%). Mp 160-164 °C (EtOAc-hexane). IR $(CHCl₃)$ cm⁻¹: 1727, 1693. ¹H NMR (300 MHz, CDCl₃) δ : 2.33 (3H, s, N–COMe), 4.32 (2H, s, COCH₂N–), 7.49

(1H, d, J=1.5 Hz, Ar-H), 8.76 (1H, brs, Ar-H). ¹³C NMR (CDCl3, 100 MHz) ^d: 24.5, 56.7, 119.5, 120.4, 121.5, 131.5, 132.1, 155.0, 167.8, 190.7. MS m/z (%): 335 (M+4, 23), 333 $(M+2, 47)$, 331 $(M⁺, 24)$, 293 (48), 291 (100), 289 (51), 263 (23), 43 (25). HRMS m/z calcd for $C_{10}H_7Br_2NO_2$: 330.8844. Found: 330.8836.

4.5.3. 3-Allyl-4,6-dibromo-3-hydroxyindolin-2-one (16). A solution of bromine (1.0 M in CH₂Cl₂, 78 μ l) was added to a solution of 14 (7.8 mg, 23.5 μ mol) in CH₂Cl₂ at 0 °C for 2 h. The reaction mixture was extracted with EtOAc, and the extract was then washed with sat. NaHCO₃ and brine, dried over MgSO4, and concentrated under reduce pressure to give a residue. A mixture of the residue (9.7 mg), allyl alcohol (d=0.85, 6.2 μ l, 91 μ mol) and MS 4A (22 mg) in DMF (1 ml) was stirred at room temperature for 2 days. The reaction mixture was diluted with diethyl ether and filtrated through Celite[®] 545. The filtrate was evaporated under reduced pressure to give a residue, which was extracted with EtOAc and 5% NH4OH. The organic layer was washed with brine, dried over MgSO4, and concentrated under reduced pressure. The obtained residue (9.1 mg) and DBU (3.6 mg, 23.5 μ mol) in toluene (1 ml) was stirred at 40 °C for 3 h. The mixture was neutralized with AcOH, diluted in H_2O , and extracted with EtOAc. The extract was washed with brine, dried over $MgSO₄$, and concentrated under reduced pressure. A mixture of the residue and 10% LiOH (25 μ l) in MeOH was stirred at room temperature for 3 days. After evaporation, an EtOAc solution of the residue was washed with H_2O and brine, dried over $MgSO_4$, and concentrated. The residue was purified by silica gel column chromatography with EtOAc–hexane (1:1) as an eluent to give indolin-2-one 16 (6.9 mg, 84%). Mp 220–222 °C (EtOAc– hexane). IR (KBr) cm⁻¹: 3378, 1733, 1641. ¹H NMR (300 MHz, acetone- d_6) δ : 2.59 (1H, dd, J=12.8, 7.4 Hz, $-CH_2$ –), 3.08 (1H, dd, J=12.8, 7.4 Hz, –CH₂–), 4.77 (1H, ddt, J=10.1, 2.2, 1.3 Hz, $=CH_2$), 4.91 (1H, ddt, J=17.3, 2.2, 1.3 Hz, $=CH_2$), 5.09 (1H, s, –OH), 5.22 (1H, ddt, $J=17.3$, 10.1, 7.4 Hz, $-CH=CH_2$), 6.94 (1H, d, $J=1.7$ Hz, Ar-H), 7.22 (1H, d, J=1.7 Hz, Ar-H). MS m/z (%): 349 $(M+4, 2), 347 (M+2, 4), 345 (M⁺, 2), 308 (49), 306 (100),$ 304 (51). HRMS m/z calcd for $C_{11}H_9Br_2NO_2$: 344.9000. Found: 344.9003.

4.5.4. (\pm) -Convolutamydine E (2b). A solution of 3allylindolin-2-one 16 (10 mg, 29 μ mol), OsO₄ (4% aqueous solution, $d=1.04$, 53 μ l, 8.7 μ mol), and NMO (50%) aqueous solution, $d=1.13$, 13 μ l, 63 μ mol) in MeCN (3 ml) was stirred at room temperature for 1 h. The reaction mixture was evaporated under reduced pressure. A solution of the residue and NaIO₄ (7.5 mg, 30 μ mol) in aqueous 1,4dioxan (3 ml) was stirred at room temperature for 1 h. The reaction mixture was extracted with EtOAc, and the extract was then washed with H₂O and brine, dried over MgSO₄, and concentrated under reduced pressure. To a solution of the residue in MeOH (2 ml) , NaBH₄ $(11 \text{ mg}, 0.3 \text{ mmol})$ was gradually added at 0° C, and the mixture was then stirred at 0° C for 30 min. After adding aqueous NH₄Cl followed by evaporating the MeOH under reduced pressure, the residue was extracted with EtOAc. The organic layer was washed with H_2O and brine, dried over $MgSO_4$, and concentrated under reduced pressure. The residue was purified by silica gel column chromatography with EtOAc–hexane (3:1) as

an eluent to give (\pm) -convolutamydine E (2b, 6.6 mg, 65%). Mp 204-206 °C (CHCl₃) [lit.^{[2c](#page-9-0)} oil]. IR (KBr) cm⁻¹: 3350, 1736, 1605, 1261, 1088, 801. ¹ H NMR (300 MHz, C₅D₅N) δ : 3.10 (2H, t, J=7.3 Hz, -C–CH₂CH₂O–), 3.96 $(2H, m, -C-CH_2CH_2O-), 7.02$ (1H, d, $J=1.6$ Hz, Ar-H), 7.41 (1H, d, J=1.6 Hz, Ar-H), 8.31 (1H, s, -NH). ¹³C NMR $(C_5D_5N, 125 MHz)$ δ : 39.3, 58.1, 77.4, 112.6, 112.6, 120.9, 128.2, 130.2, 146.7, 180.6. MS (FAB) m/z (%): 354 $(MH⁺+4, 50), 352 (MH⁺+2, 99), 350 (MH⁺, 53), 336$ (20), 334 (32), 332 (16), 306 (57), 304 (100), 302 (50). HRMS (FAB, MH⁺) m/z calcd for $C_{10}H_9Br_2NO_3$: 349.9027. Found: 349.9031.

4.5.5. 3-Allyl-4,6-dibromo-3-(tert-butyldimethylsilyloxy)indolin-2-one (17). A solution of 3-hydroxyindolin-2 one 16 (59 mg, 0.17 mmol), TBDMSOTf (235 mg, 0.89 mmol), and 2,6-lutidin (95 mg, 0.89 mmol) in CH_2Cl_2 was stirred at 0 °C to room temperature. After 3 days, the reaction mixture was quenched with brine and extracted with CH_2Cl_2 . The extract was dried over $MgSO_4$ and concentrated under reduced pressure. A solution of the residue in AcOH–THF–H₂O $(1:1:1, 1 \text{ ml})$ was stirred at 80 °C for 2 h. After cooling, the reaction mixture was extracted with EtOAc. The extract was washed with H_2O and brine, dried over MgSO4, and concentrated under reduced pressure. The residue was subjected to chromatography on a silica gel column with EtOAc–hexane (1:4) as an eluent to give 3-silyloxy-indolin-2-one 17 (56 mg, 69%). Viscous oil. IR (CHCl₃) cm⁻¹: 3427, 1732, 1641. ¹H NMR (300 MHz, CDCl₃) δ : -0.13 (3H, s, -SiMe), 0.06 (3H, s, $-SiMe$), 0.90 (9H, s, $-Si^tBu$), 2.78 (1H, dd, J=12.8, 7.5 Hz, $=CH-CH-$), 3.24 (1H, dd, J=12.8, 7.0 Hz, $=CH-CH-$), 4.92 (1H, d, J=9.9 Hz, C=CH₂), 5.17 (1H, d, J=16.5 Hz, $C=CH_2$), 5.29 (1H, m, $-CH=CH_2$), 6.97 (1H, d, $J=1.4$ Hz, Ar-H), 7.34 (1H, d, $J=1.4$ Hz, Ar-H), 8.25 (1H, br, NH). ¹³C NMR (CDCl₃, 100 MHz) δ : -3.5, -3.4, 18.4, 25.8, 40.7, 79.3, 112.3, 120.0, 120.5, 123.2, 127.8, 129.4, 129.7, 142.6, 177.3. MS (FAB) m/z (%): 464 (MH⁺+4, 4), 462 (MH⁺+2, 7), 460 (MH⁺, 5), 4221 (7), 420 (10), 4187 (6), 4065 (15), 404 (24), 402 (12), 365 (12), 363 23), 361 12), 332 (51), 330 (100), 328 (52). HRMS (FAB, MH⁺) m/z calcd for $C_{17}H_{23}Br_2NO_2Si$: 459.9943. Found: 459.9920.

4.5.6. 4,6-Dibromo-3-(tert-butyldimethylsilyloxy)-3-(2 oxopropyl)indolin-2-one (18). A suspension of palladium (II) chloride (3.2 mg, 0. 019 mmol) and copper (I) chloride $(18.8 \text{ mg}, 0.19 \text{ mmol})$ in dioxane–H₂O $(7:1, 3 \text{ ml})$ was vigorously stirred with bubbling oxygen gas at room temperature for 1 h under oxygen atmosphere. The indolin-2-one 17 (87 mg, 0.19 mmol) was added to the mixture. After heating at 50° C for 24 h, the reaction mixture was cooled to room temperature and diluted with $CH₂Cl₂$. The precipitate was filtered off, and the filtrate was dried over $MgSO₄$, and evaporated off. The obtained residue was purified by flash silica gel column chromatography with EtOAc–hexane $(1:1)$ as an eluent to give 3-(tert-butyldimethylsilyloxy)-3-(2-oxopropyl)indolin-2-one (18, 11 mg, 12%). Viscous oil. IR (CHCl₃) cm⁻¹: 3431, 1747, 1718. ¹H NMR (300 MHz, CDCl₃) δ : -0.21 (3H, s, $-SiMe$), 0.04 (3H, s, –SiMe), 0.87 (9H, s, –Si^t Bu), 2.08 (3H, s, –COMe), 3.31 $(1H, d, J=18.2 \text{ Hz}, -CHCO-), 3.99 (1H, d, J=18.2 \text{ Hz},$ -CHCO-), 6.97 (1H, d, J=1.4 Hz, Ar-H), 7.27 (1H, d, $J=1.4$ Hz, Ar-H), 7.56 (1H, br, $-NH$). MS (FAB) m/z (%): 480

 $(MH⁺+4, 14)$, 478 $(MH⁺+2, 26)$, 476 $(MH⁺, 13)$, 422 (12), 420 (23), 418 12), 348 (16), 346 (30), 344 (16), 306 (50), 34 (100), 302 (34). HRMS (FAB, MH⁺) m/z calcd for C₁₇H₂₃-Br2NO3Si: 475.9892. Found: 475.9879.

4.5.7. (\pm) -Convolutamydine A (2a). Tris(dimethylamino)sulfur (trimethylsilyl)difluoride (TAS-F) (6.4 mg, 23 μ mol) was gradually added to a stirred solution of indolin-2-one 17 (11 mg, 23 μ mol) in dry DMF (1 ml) at 0 °C. The mixture was stirred at 15 \degree C for 2.5 h, diluted with EtOAc, washed with sat. $KHSO₄$, and extracted with EtOAc. The extract was dried over MgSO₄ and evaporated. The obtained residue was purified by silica gel column chromatography with EtOAc–hexane (3:2) as an eluent to give (\pm) convolutamydine A (2a, 6.3 mg, 75%). Mp $192-195$ °C (EtOAc–hexane) [lit.^{2b} mp 190–195 °C]. IR (CHCl₃) cm⁻¹: 3430, 1747, 1610. ¹H NMR (300 MHz, CDCl₃) δ : 2.16 (3H, s, –COMe), 3.34 (1H, d, $J=17.3$ Hz, –CHCO–), 3.73 (1H, d, $J=17.3$ Hz, $-CHCO-$), 5.12 (1H, br, $-OH$), 7.00 (1H, d, $J=1.6$ Hz, Ar-H), 7.32 (1H, d, $J=1.7$ Hz, Ar-H), 7.74 (1H, s, $-NH$). MS m/z (%): 365 (M+4, 20), 363 $(M+2, 42)$, 361 $(M⁺, 21)$, 322 (8) , 320 (16) , 318 (10) , 308 (47), 306 (100), 304 (59), 279 (44), 277 (91), 275 (48), 252 (14), 250 (31), 248 (18), 170 (18), 168 (18), 88 (12). HRMS m/z calcd for $C_{11}H_9Br_2NO_3$: 360.8949. Found: 360.8947.

Acknowledgements

We wish to thank N. Eguchi and T. Koseki, S. Kubota, and T. Suzuki in the Analytical Center of our University for conducting microanalysis and obtaining mass spectra. This work was financially supported by a Grant-in-Aid (No. 14572018) for Scientific Research (C) from the Ministry of Education, Science, Sports, and Culture, Japan.

References and notes

- 1. Isolation: (a) Ubaidullaev, K. A.; Shakirov, R.; Yunosov, S. Y. Khim. Prir. Soedin. 1976, 12, 553–554. Synthesis: (b) Rasmussen, H. B.; MacLeod, J. K. J. Nat. Prod. 1997, 60, 1152–1154.
- 2. Isolation: (a) Kamano, Y.; Zhang, H.-P.; Ichihara, Y.; Kizu, H.; Komiyama, K.; Pettit, G. R. Tetrahedron Lett. 1995, 36, 2783–2784. (b) Zhang, H.-P.; Kamano, Y.; Ichihara, Y.; Kizu, H.; Komiyama, K.; Itokawa, H.; Pettit, G. R. Tetrahedron 1995, 51, 5523–5528. (c) Kamano, Y.; Kotake, A.; Hashima, H.; Hayakawa, I.; Hiraide, H.; Zhang, H.-P.; Kizu, H.; Komiyama, K.; Hayashi, M.; Pettit, G. R. Collect. Czech. Chem. Commun. 1999, 64, 1147–1153. Synthesis: (d) Garden, S. J.; Torres, J. C.; Ferreira, A. A.; Silva, R. B.; Pinto, A. C. Tetrahedron Lett. 1997, 38, 1501–1504. (e) Miah, S.; Moody, C. J.; Richards, I. C.; Slawin, A. M. Z. J. Chem. Soc., Perkin Trans. 1 1997, 2405–2412. (f) Jnaneshwara, G. K.; Bedekar, A. V.; Deshpande, V. H. Synth. Commun. 1999, 29, 3627–3633. (g) Jnaneshwar, G. K.; Deshpande, V. H. J. Chem. Res. (S) 1999, 632–633.
- 3. Isolation: (a) Monde, K.; Sasaki, K.; Shirata, A.; Takasugi, M. Phytochemistry 1991, 30, 2915–2917. Synthesis: (b) Suchy, M.; Kutschy, P.; Monde, K.; Goto, H.; Harada, N.; Takasugi,

M.; Dzurilla, M.; Balentova, E. J. Org. Chem. 2001, 66, 3940–3947.

- 4. Jimenez, J. I.; Huber, U.; Moore, R. E.; Patterson, G. M. L. J. Nat. Prod. 1999, 62, 569–572.
- 5. Fréchard, A.; Fabre, N.; Péan, C.; Montaut, S.; Fauvel, M.-T.; Rollin, P.; Fourasté, I. Tetrahedron Lett. 2001, 42, 9015–9017.
- 6. Isolation: (a) Koguchi, Y.; Kohno, J.; Nishio, M.; Takahashi, K.; Okuda, T.; Ohnuki, T.; Komatsubara, S. J. Antibiot. 2000, 53, 105–109. Structure: (b) Khono, J.; Koguchi, Y.; Nishio, M.; Nakao, K.; Juroda, M.; Shimizu, R.; Ohnuki, T.; Komatsubara, S. J. Org. Chem. 2000, 65, 990–995. Synthesis: (c) Lin, S.; Danishefsky, S. J. Angew. Chem., Int. Ed. 2002, 41, 512–515. (d) Albrecht, B. K.; Williams, R. M. Org. Lett. 2003, 5, 197–200.
- 7. Others: (a) Pedras, M. S.; Montaut, S.; Xu, Y.; Khan, A. Q.; Loukaci, A. Chem. Commun. 2001, 1572–1573. (b) Hewawasam, P.; Erway, M.; Moon, S. L.; Knipe, J.; Weiner, H.; Boissard, C. G.; Post-Munson, D. J.; Gao, Q.; Huang, S.; Gribkoff, V. K.; Meanwell, N. A. J. Med. Chem. 2002, 45, 1487–1499. (c) Codding, P. W.; Lee, T. A.; Richardson, J. F. J. Med. Chem. 1984, 27, 649–654. (d) Popp, F. D.; Pajouhesh, H. J. Pharm. Sci. 1982, 71, 1052–1054.
- 8. Sundberg, R. J. The chemistry of indoles; Academic: New York, 1970; pp 341–392.
- 9. Aldol condensation of isatins with methylketones: (a) Garden, S. J.; da Silva, R. B.; Pinto, A. C. Tetrahedron 2002, 58, 8399–8412. (b) Beccalli, E. M.; Marchesini, A.; Pilati, T. J. Chem. Soc., Perkin Trans. 1 1994, 579–587. (c) Pajouhesh, H.; Parson, R.; Popp, F. D. J. Pharm. Sci. 1983, 72, 318–321. (d) Popp, F. D. J. Heterocycl. Chem. 1982, 19, 589–592.
- 10. Addition of Grignard reagent to isatins: (a) Sharma, V. M.; Prasanna, P.; Seshu, K. V. A.; Renuka, B.; Rao, C. V. L.; Kumar, G. S.; Narasimhulu, C. P.; Babu, P. A.; Puranik, R. C.; Subramanyam, D.; Venkateswarlu, A.; Rajagopal, S.; Kumar, K. B. S.; Rao, C. S.; Mamidi, N. V. S. R.; Deevi, D. S.; Ajaykumar, R.; Rajagopalan, R. Bioorg. Med. Chem. Lett. 2002, 12, 2303–2307. (b) Hewawasam, P.; Gribkoff, V. K.; Pendri, Y.; Dworetzky, S. I.; Meanwell, N. A.; Martinez, E.; Boissard, C. G.; Post-Munson, D. J.; Trojnacki, J. T.; Yeleswaram, K.; Pajor, L. M.; Knipe, J.; Gao, Q.; Perrone, R.; Starrett, J. E. Bioorg. Med. Chem. Lett. 2002, 12, 1023–1026. (c) Hewawasam, P.; Erway, M. Tetrahedron Lett. 1998, 39, 3981–3984.
- 11. Baylis-Hilman reaction of isatins with enones: Garden, S. J.; Skakle, J. M. S. Tetrahedron Lett. 2002, 43, 1969–1972.
- 12. Photoinduced addition of enol ethers to isatin: Zhang, Y.; Xue, J.; Gao, Y.; Fun, H.-K.; Xu, J.-H. J. Chem. Soc., Perkin Trans. 1 2002, 345–353.
- 13. Oxidation of 3-substituted indoles: (a) Peyrot, F.; Martin, M.-T.; Migault, J.; Ducrocq, C. Eur. J. Org. Chem. 2003, 172–181. (b) Alvarez, R. G.; Hunter, I. S.; Suckling, C. J.; Thomas, M.; Vitinius, U. Tetrahedron 2001, 57, 8581–8587.
- 14. Oxidation of 3-substituted indolin-2-ones: (a) Takayama, H.; Shimizu, T.; Sada, H.; Harada, Y.; Kitajima, M.; Aimi, N. Tetrahedron 1999, 55, 6841–6846. (b) Miah, S.; Moody, C. J.; Richards, I. C.; Slawin, A. M. Z. J. Chem. Soc., Perkin Trans. 1 1997, 2405–2412. (c) Hewawasam, P.; Meanwell, N. A.; Gribkoff, V. K.; Dworetzky, S. I.; Boissard, C. G. Bioorg. Med. Chem. Lett. 1997, 7, 1255–1260.
- 15. Oxidation of 3-alkylideneindolin-2-ones: (a) Lin, S.; Danishefsky, S. J. Angew. Chem., Int. Ed. 2002, 41, 512–515. (b) Lin, S.; Danishefsky, S. J. Angew. Chem., Int.

Ed. 2001, 40, 1967–1970. (c) Albrecht, B. K.; Williams, R. M. Org. Lett. 2003, 5, 197–200. (d) Inoue, M.; Furuyama, H.; Sakazaki, H.; Hirama, M. Org. Lett. 2001, 3, 2863–2865.

- 16. Carbonylation of litthiated N-pivaloyanilines: Smith, K.; El-Hiti, G. A.; Pritchard, G. J.; Hamilton, A. J. Chem. Soc., Perkin Trans. 1 1999, 2299–2303.
- 17. Miscellaneous reactions: (a) Nicolaou, K. C.; Rao, P. B.; Hao, J.; Reddy, M. V.; Rassias, G.; Huang, X.; Chen, D. Y.-K.; Snyder, S. A. Angew. Chem., Int. Ed. 2003, 42, 1753–1758. (b) Pushechnikov, A. O.; Volochnyuk, D. M.; Tolmachev, A. A. Synlett 2002, 1140-1142. (c) Kafka, S.; Klásek, A.; Kosmrlj, J. J. Org. Chem. 2001, 66, 6394–6399. (d) Klásek, A.; Koistek, K.; Polis, J.; Kosmrlj, J. Tetrahedron 2000, 56, 1551–1560. (e) Dallacker, F.; Sanders, G. Chemiker-Zeitung 1986, 110, 405–411. (f) Bogavac, M.; Arsenijevic, L.; Pavlov, S.; Arsenijevic, V. Arh. Farm. 1985, 35, 99–103.
- 18. (a) Nair, V.; Jayan, C. N. Tetrahedron Lett. 2000, 41, 1091–1094. (b) Nair, V.; Jayan, C. N.; Ros, S. Tetrahedron 2001, 57, 9453–9459.
- 19. Nair, V.; Ros, S.; Jayan, C. N. J. Chem. Res. (S) 2001, 551–553.
- 20. Mikhailovskii, A. G.; Ignatenko, A. V.; Bubnov, Yu. N. Chem. Heterocycl. Compd. (USSR) (Engl. Transl.) 1998, 34, 785–790.
- 21. Malapel-Andrieu, B.; Piroelle, S.; Merour, J.-Y. J. Chem. Res. (S) 1998, 594–595.
- 22. (a) Kawasaki, T.; Masuda, K.; Baba, Y.; Takada, K.; Sakamoto, M. Chem. Pharm. Bull. 1994, 42, 1974–1976. (b) Kawasaki, T.; Masuda, K.; Baba, Y.; Terashima, R.; Takada, K.; Sakamoto, M. J. Chem. Soc., Perkin Trans. 1 1996, 729–733.
- 23. (a) Kawasaki, T.; Terashima, R.; Sakaguchi, K.; Sekiguchi, H.; Sakamoto, M. Tetrahedron Lett. 1996, 37, 7525–7528. (b) Kawasaki, T.; Ogawa, A.; Takashima, Y.; Sakamoto, M. Tetrahedron Lett. 2003, 44, 1591–1593.
- 24. (a) Kawasaki, T.; Ohtsuka, H.; Sakamoto, M. J. Chem. Soc., Chem. Commun. 1990, 781–782. (b) Kawasaki, T.; Ohtsuka, H.; Mihira, A.; Sakamoto, M. Heterocycles 1998, 47, 367–373.
- 25. Wipf, P. Comprehensive organic synthesis; Trost, B. M., Ed.; Pergamon: Oxford, 1991; Vol. 5, pp 827–873.
- 26. Kafka, S.; Klásek, A.; Kosmrlj, J. J. Org. Chem. 2001, 66, 6394–6399.
- 27. (a) Tsuji, J. Synthesis 1984, 369–384. (b) Yatagai, H.; Yamanoto, Y.; Maruyama, K. J. Am. Chem. Soc. 1980, 102, 4548–4550. (c) Nagashima, H.; Sakai, K.; Tsuji, J. Chem. Lett. 1982, 859-860. (d) Tsuji, J.; Nagashima, H.; Hori, K. Tetrahedron Lett. 1982, 23, 2679–2682.
- 28. Desilylation of TBDMS ether 10 with TBAF in the absence of AcOH gave the desired hydroxy derivative 11 (20%) and isatine (50%), which was caused by retro-aldol reaction of 11 under basic conditions.
- 29. Martin, P. Helv. Chim. Acta 1988, 71, 344-347.
- 30. Kawasaki, T.; Nonaka, Y.; Matsumura, K.; Monai, M.; Sakamoto, M. Synth. Commun. 1999, 29, 3251–3261.
- 31. The results of Wacker oxidation of the TBDMS ether of 17 having bromine atoms were poor, giving only a 12% yield of the desired acetonyl derivative.
- 32. Scheidt, K. A.; Chen, H.; Follows, B. C.; Chemler, S. R.; Coffey, D. S.; Roush, W. R. J. Org. Chem. 1998, 63, 6436–6437.